

OCCURRENCE OF C₈ UNITS IN NATURALLY OCCURRING STILBENES AND PHENYL ISOCOUMARINS

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THE simplest of naturally occurring stilbenes and the earliest to be discovered was pinosylvin.1 This substance, its methyl ethers and dihydro-derivatives are commonly found in the heartwoods of pine trees and one or more of Similarly these are invariable components. pterostilbene is a characteristic component of the wood of Pterocarpus species.2-4 The occurrence of these compounds has therefore been considered to be of importance for the taxonomy of these plants. They seem to play an important part in protecting the wood from insect and fungal attack. Pterostilbene has also been found to have a curative effect on diabetes. More recently other stilbenes have been isolated mainly from the heartwood of timber-yielding plants and from certain plant drugs. The chief members occurring in Nature are given in the following table (Table I).

In naturally-occurring stilbenes the substitution of the benzene rings is of a special nature. The majority of them have no hydroxyl in an ortho-position. All of them have the 3:5-dihydroxy (methoxy) substitution in one of the rings and suggest derivation from an orcinol (or orsellinic, C_8) unit. The second part has mostly the normal substitution of the C_9 unit of eugenol derivatives, lignanes and flavonoids, i.e., 4', 3':4' and 3':4':5'. The 2':4' substitution should also be taken as normal variation, though somewhat rare (Rao and Seshadri⁵). Then the following scheme for the formation of stilbenes could be suggested involving a derived C_8 unit (Seshadri⁶) and a derived C_9 unit.

OH (DOPA;
$$C_{9}$$
.)
OH (OH)
$$CHO$$

	Name	"	Structure	Source
1 2	Pinosylvin Pinosylvin monomethyl etl	her	3:5-dihydroxy 3-hydroxy-5-methoxy	Heartwood of Pinus sylvestris, and P. nigra Heartwood of Pinus strobus, P. cembra and P. montana
3 4 5 6 7	Pinosylvin dimethyl ether Resveratrol Pterostilbene Hydroxy resveratrol	•••	3:5:dimethoxy 3:5:4'-trihydroxy 3:5-dimethoxy-4'-hydroxy 3:5:2':4'-tetrahydroxy 3:5:3':4'-tetrahydroxy	Heartwood of P. nigra and P. palustris Root of Veratrum grandistorum Heartwood of Pterocarpus species Root of Veratrum grandistorum Heartwood of Vouacapous macropetala and V. americana
8 9 10	Rhapontigenin Chlorophorin	••	3:5:3':4':5'-pentahydroxy 3:5:3'-trihydroxy-4'-methoxy 3:5:2':4'-tetrahydroxy 4-(3:7-dimethyl octa-2:6-dienyl)	Heartwood of Vouacapous macropetala Root of Rheum emodi Heartwood of Chlorophora excelsa

The above scheme seems to be supported by two observations: (i) the occurrence of stilbene-o-carboxylic acids and 3-phenyl dihydroisocoumarins in nature. Two related 3-phenyldihydro-isocoumarins, hydrangenol (III a) and phyllodulcinol (IIIb) are present in varieties of Hydrangea opuloides. They are accompanied by the corresponding stilbene-o-carboxylic acids called iso-hydrangenol (IV a) and isophyllodulcinol (IV b). These compounds have been fully investigated by Asahina and Asano.7 The dihydro-isocoumarins can by treatment with alkali be readily converted into the stilbene carboxylic acids and the latter on heating close up the ring to yield the former.

It could be noticed that they have lost one of the two meta di-hydroxyl groups found in formulæ (I and II). The stability of these acids natural on the above scheme of biogenesis if the substitution (Va to Vb) should take place prior to the final decarboxylation of the acid, since carbonyl derivatives of resorcinol all undergo alkylations in the 3-position which corresponds to the 4-position of stilbenes.

The question of the biogenesis of naturally occurring stilbenes has received some attention earlier. Geissman and Hinreiner⁹ attempted to adopt the $C_6 + C_9$ scheme as found in anthoxanthins and have focussed attention on the difficulties of the explanation. Sir Robert Robinson¹⁰ has derived the structures of pinosylvin, pterostilbene and chlorophorin from polyketomethylenes; the scheme is quite suitable for these compounds but may not suit the other stilbenes having the 3':4'-substitution pattern of C_9 units.

could be attributed to this event which should have happened earlier by a process of reduction; the isocompounds (acids) may thus have been enabled to escape decarboxylation and undergo ring closure to yield the isocoumarins. The presence of a resorcylic acid structure would lead to ready decarboxylation, yielding the stilbenes.

(ii) The structure of chlorophorin's (VI). The special feature of chlorophorin is the di-isoprenoid side chain that is found in the 4-position located between two phenolic hydroxyl groups. It would naturally be an inaccessible position if the substitution (or entry of the long chain) should be considered to take place in a preformed stilbene structure. But it becomes quite

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