## SHORT COMMUNICATIONS

CRYSTAL DATA ON LITHIUM p-NITRO-BENZOATE, LITHIUM 2,4-DINITRO-PHENOLATE(1,10-PHENANTHROLINE), LITHIUM PICRATE(1,10-PHENANTHROLINE) AND LITHIUM PICRATE(18-CROWN-6)

SUBHASH PRABHAKAR, V. W. BHAGWAT\*, V. VIJAYVARGIYA and N. S. POONIA

Department of Chemistry, University of Indore, Indore 452001, India.

SODIUM, potassium, magnesium and calcium play intriguing roles in biology<sup>1,2</sup> which are difficult to understand in view of the existing knowledge<sup>3</sup> about the interaction chemistry of s-block cations (abbr. M<sup>z+</sup>) in general. This has led us to undertake broadbased studies<sup>4</sup> on M<sup>z+</sup> ions which include synthesis and characterization of M<sup>z+</sup>-compounds involving different anionic and neutral nucleophiles. The main point of examination is whether the product is a genuine homogenous solid phase and whether the

M<sup>z+</sup>-nucleophile interactions are definite. Herein we report (table 1) preliminary cell data on four lithium-derivatives all of which are novel crystalline phases. Cell data were calculated from oscillation and Weissenberg photographs with CuKα radiation. Density measurements were made by the flotation technique using benzene/bromoform mixtures.

The authors thank the DST (New Delhi) for financing this project. SP and VWB thank the UGC for financial support.

17 May 1984; Revised 27 August 1984

Table 1 Crystal data on some lithium derivativesa

	LiPnb	LiDnp(PHEN)	LiPic(PHEN)	LiPic(18-C-6)
Crystal	White, semi- transparent, needles	Dark yellow, transparent	Yellow transparent	Light yellow, trans- parent needles
Mol. Formula	Li(C <sub>7</sub> O <sub>4</sub> NH <sub>4</sub> )· 3H <sub>2</sub> O	$Li(C_6N_2O_5H_3)$ $(C_{12}N_2H_8)$	$Li(C_6N_3O_7H_2)$ $(C_{12}N_2H_8)$	$Li(C_6N_3O_7H_2)$ $(C_{12}O_6H_{24})$
Crystal system  a b c $\alpha$ $\beta$ $\gamma$ $\gamma$	Orthorhombic 7.11 ± 0.07 Å 26.14 ± 0.04 Å 11.22 ± 0.05 Å	Triclinic $11.22 \pm 0.04 \text{ Å}$ $22.28 \pm 0.07 \text{ Å}$ $13.65 \pm 0.07 \text{ Å}$ $85.7^{\circ} \pm 0.3^{\circ}$ $92.2 \pm 0.7$ $94.5 \pm 0.4$ $3215.3 \text{ Å}^{3}$	Triclinic $17.93 \pm 0.03 \text{ Å}$ $12.46 \pm 0.05 \text{ Å}$ $15.65 \pm 0.06 \text{ Å}$ $75.2^{\circ} \pm 0.7^{\circ}$ $113.9 \pm 0.4$ $126.8 \pm 0.4$ $2474.6 \text{ Å}^{3}$	Triclinic $7.84 \pm 0.07 \text{ Å}$ $13.22 \pm 0.05 \text{ Å}$ $15.78 \pm 0.04 \text{ Å}$ $96.9^{\circ} \pm 0.8^{\circ}$ $87.25 \pm 0.6$ $95.8 \pm 0.6$ $1551.9 \text{ Å}^{3}$
Mol. Weight Density (Cal) Density (Exp) "CuKa Z Space Group	227.01 1.433 gm/cc 1.419 1.17 cm <sup>-1</sup> 8 P <sub>2,2,2,1</sub>	370.12 1.529 gm/cc 1.531 0.99 cm <sup>-1</sup> 8 P1 or PT	415.12 1.667 gm/cc 1.664 1.16 cm <sup>-1</sup> 6 P1 or PT	499.12 1.068 gm/cc 1.130* 0.88 cm <sup>-1</sup> 2 P1 or PT

<sup>&</sup>lt;sup>a</sup> Key to abbreviations:— Pnb: para-nitrobenzoate, Dnp: 2,4-dinitrophenolate, PHEN: 1,10-phenanthroline, Pic: 2,4,6-trinitrophenolate (Picrate), 18-C-6: 18-Crown-6.

<sup>\*</sup>School of Studies in Chemistry, Vikram University, Ujjain 456010, India.

<sup>1.</sup> Bittar, E. E., (ed.), Membranes and Ion Transport, Wiley-Inter Science, London, 1970.

<sup>2.</sup> Kotyk, A. and Janacek, K., Cell Membrane Transport, Plenum, London, 1970.

<sup>3.</sup> Hughes, M. N., The Inorganic Chemistry of Biological Systems, Wiley, New York, 1972.

<sup>4.</sup> Poonia, N. S. and Bajaj, A. V., Chem. Rev., 1979, 79, 389.

<sup>\*</sup> Crystals are slightly soluble in benzene/bromoform mixture.